#### Title

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### Spiral Cold Electrode Fluorescent Lamp

## Background of the Present Invention

#### Field of Invention

The p resent invention relates to a spiral cold electrode fluorescent lamp, and more particularly to a spiral electrode fluorescent lamp whose second electrode is coated with a layer of gas absorbent for slowing the decaying rate of the same.

#### **Description of Related Arts**

A compact fluorescent lamp (CFL) is widely used for lightening. A conventional CFL includes a light tube, spread with a phosphor coating on its inner surface, containing inert gas and mercury substance, in the form of mercury vapor or liquid mercury. The light tube is enclosed with caps at its two ends, at which a first and second electrodes are disposed therein. When enough electric voltage is applied to the first and second electrodes, the second electrode emits electrons and causes the mercury to discharge, thereby conducting the electric current to the first electrode. In the course of discharge, the mercury emits ultra violet rays which excite the phosphor coating to generate visible light. The second electrode is usually shaped as a wire in a dimension of millimeter. In order to electrically excite the mercury to emit ultra violet rays, the second electrode is usually required to function at a temperature about 800 degrees Celsius.

A cold electrode fluorescent lamp (CCFL) has a basic structure similar to CFL in the sense that they all need a light tube with an inner layer of phosphor coating that contains inert gas and mercury substance, and a second electrode electrically connected to a power source for exciting the mercury. The CCFL is different from the CFL in the sense that the second electrode of CCFL has a larger surface area and lower functioning temperature. The second electrode of CCFL is u sually shaped as a single or multiple layers of plates, such that its surface area is larger than the wire-shaped second electrode

of CFL. Additionally, only a temperature about 100 degrees Celsius is required for the second electrode of CCFL to function. This is how the name "cold electrode" is given, comparing the traditional second electrode for CFL. Because the cold electrode functions at a lower temperature, the life span of CCFL usually lasts longer than its comparative models of CFL. Moreover, the CCFL can better survive an impact force than the CFL does, because it is easier for the impact force to disconnect the wire-shaped second electrode of CFL from the power source than to disconnect the plate-shaped second electrode from the same.

One phenomenon causes the cold electrode to decay is its oxidation problem. Besides inert gas and mercury, the light tube always contains air either residually left in the light tube, or subsequently entered from the sealing into the same. During the manufacturing process, gases such as O<sub>2</sub>, CO, CO<sub>2</sub> and H<sub>2</sub>O, may be existed in the light-tube and such residual active gases would facilitate the oxidation of the cold electrode. The oxidation decreases the intensity of electrons emitted from the cold electrode, and therefore reducing the luminance of the CCFL. Until the oxidation develops to a certain point, the cold electrode can no longer emit electrons with enough intensity to excite the mercury. At this point, the CCFL can no longer serve its purpose of illumination.

One solution to cope with the oxidation problem is to place a gas absorber in the light tube to absorb the oxygenic gas. The less the oxygenic gas exists in the light tube, the slower the cold electrode oxidizes, the longer the cold electrode is able to emit electrons with sufficient intensity. The life span of the CCFL is therefore increased. For example, a conventional color display may adopt a second electrode partially coated with a layer of gas absorbent based on barium alloy; a filament light bulb may contain gas absorber having phosphor as its predominating constitutient; and some high-end products of CFL include gas absorber made of alloy containing zirconium and aluminum. These various types of gas absorber serve the same purpose of absorbing oxygenic gas in order to lengthen the life span of the lights.

One shortcoming of the conventional gas absorber is its insufficient capability of absorbing the oxygenic gas. The gas absorber performs usually at an activation temperature as high as 900 degrees Celsius. The high activation temperature works in both ways. Although it helps the absorber to absorb the oxygenic gas, it facilitates the oxidizing reaction of the second electrode. However, it also requires an additional expensive manufacturing equipments to activate the gas absorber at 900 degrees Celsius

to form activated gas absorber in order to absorb oxygenic gas at normal temperature. As a result, the cost of manufacturing the CCFL is inevitably increased and processing of making the same is likewise complicated.

Thus, what is needed is a CCFL containing a gas absorber in its light tube that has an improved capability of absorbing oxygenic gas and lower requirement for working temperature in order to lengthen the life span of the second electrode of CCFL.

## Summary of the Present Invention

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An objective of the present invention is to provide a light tube for cold electrode fluorescent lamp that includes a zirconium-vanadium-iron-based gas absorber, which is able to be activated at an activation temperature substantially lower than 900 degrees Celsius and has better efficiency of gas absorption than conventional ones.

Another objective of the present invention is to provide a light tube for cold electrode fluorescent lamp that includes a zirconium-vanadium-iron-based gas absorber, which is able to be activated at an activation temperature about 390 degrees Celsius, that is during the general gas discharging step of its manufacturing process, so as to minimize its manufacturing processes and cost and to eliminate those expensive heating equipments.

Another objective of the present invention is provide a light tube for cold electrode fluorescent lamp whose second electrode is made in various shapes for enlarging a surface area of the second electrode in order to enhance the second electrode in terms of resisting oxidation and surviving an impact forced applied to the light tube.

Another objective of the present invention is to provide a light tube for cold electrode fluorescent lamp that is made in various shapes in order to reduce a space occupied by the same.

Another objective of the present invention is to provide a cold electrode fluorescent lamp that includes a housing air-tightly attached to an igniter casing extended from a base for maintaining heat therein in order to warm the second electrode.

Another objective of the present invention is to provide a cold electrode fluorescent lamp that includes a housing attached to an igniter casing extended from a base, wherein an air passage is formed between the housing and the igniter casing for balancing pressure within and without the housing.

Another objective of the present invention is to provide a cold electrode fluorescent lamp that includes an igniter for driving the cold electrode to a functioning stage.

Another objective of the present invention is to provide a cold electrode fluorescent lamp that includes a housing made of colors for effects of colorful illumination.

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In order to accomplish the above objectives, the present invention provides a light tube for a cold electrode fluorescent lamp comprising: a light tube body, having a first end portion and second end portion, containing inert gas, mercury substance and a layer of p hosphor coating on an inner surface of the light tube body; a first electrode disposed at the first end portion in the light tube body and a second electrode disposed at the second end portion in the light tube body, wherein the first and second electrodes are adapted for connecting to terminals of high frequency alternate current (AC) electric power source for emitting electrons to excite the mercury substance for conducting the electrons to the other electrode as a electric loop, wherein the excited mercury substance emits ultra violet rays causing the phosphor coating to generate visible light; and a gas absorber, made of zirconium-vanadium-iron alloy, formed at the second electrode for absorbing oxygenic gas.

These and other objectives, features, and advantages of the present invention will become apparent from the following detailed description, the accompanying drawings, and the appended claims.

## Brief Description of the Drawings

- FIG. 1 is a schematic view of a CCFL light bulb that includes an improved gas absorber in its light tube according to a preferred embodiment of the present invention.
- FIG. 2 is a perspective view of the second electrode of CCFL according to the preferred embodiment of the present invention.
  - FIG. 3 is a perspective view of an alternative of second electrode according to the preferred embodiment of the present invention.
  - FIG. 4 is a perspective view of another alternative of second electrode according to the preferred embodiment of the present invention.
- FIG. 5 is a perspective view of another alternative of second electrode according to the preferred embodiment of the present invention.
  - FIG. 6 is a perspective view of another alternative of second electrode according to the preferred embodiment of the present invention.
- FIG. 7 is a perspective view of another alternative of second electrode according to the preferred embodiment of the present invention.
  - FIG. 8 is a perspective view of the light tube of CCFL according to the preferred embodiment of the present invention.
  - FIG. 9 is a perspective view of an alternative of light tube according to the preferred embodiment of the present invention.
- FIG. 10 is a perspective view of an alternative of light tube according to the preferred embodiment of the present invention.
  - FIG. 11 is a perspective view of an alternative of light tube according to the preferred embodiment of the present invention.

# Detailed Description of the Preferred Embodiment

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Referring to FIG. 1, a CCFL light bulb that includes a gas absorber is illustrated according to a preferred embodiment of the present invention. The light bulb comprises a base 8, an igniter casing 7 extended from the base 8, an igniter 6 disposed in the igniter casing 7, electrodes including a first electrode 3a and a second electrode 3, a gas absorber agent 4 and 4a, light tube 2 containing the electrodes 3, 3a and gas absorbers 4, 4a, and a housing 1 attached to the igniter casing 7 for enclosing the light tube 2 therein.

The base 8 has a threaded side wall 81, having a cylindrical shape, and a tapered bottom end 82, adapted for securing to a compatible socket for electrically connecting to an alternate current (AC) electric power source. The threaded side wall 81 and the tapered end 82 are made of conductive material for electrically connecting to the socket. The tapered end 82 and the side wall 81 are so insulated that they may electrically be connected to terminals of the socket respectively.

The igniter casing 7 extends from the base 8, in which a cavity is formed for receiving various components. The igniter casing 7 is made integrally with the base 8 for ease of manufacturing. It is noted that the casing 7 may also be made separately from the base 8, and then attached thereto via traditional connection means.

The igniter 6, which is disposed in the igniter casing 7 for transforming voltage to a sufficient level to drive the first and second electrodes 3a, 3 to function, is electrically connected to the tapered end 82 and the side wall 81 of the base 8 through first wires 9a and 9, respectively. When the base 8 is secured on the socket, the igniter 6 may be triggered upon a turned-on switch to tap the igniter 6 to the electric power source. The igniter 6 is further electrically connected to the first and second electrodes 3a, 3 by second wires 5 and 5a extending into the light tube 3 and 3a.

The light tube 2 has a spiral shape with two end portions horizontally extending toward the igniter casing 7. The spiral shape minimizes the space the light tube 2 occupies so that the CCFL light bulb can be made compact. The light tube 2, having a phosphor coating spread on its inner surface, contains inert gas, such as neon, argon, and mercury substance. The mercury substance may be in various forms, such as mercury vapor, liquid mercury and amalgam.

The first and second electrodes 3a and 3, disposed at the end portions of the light tube 2, are electrically connected to the terminals of the socket via the second wires 5 and 5a, the igniter 6, the first wires 9 and 9a, and the side wall 81 and tapered end 82 of the base 8. Particularly, the first and second electrodes 3a and 3 are electrically connected to the high frequency AC electric power source, to from an electric loop. When the loop is switched on, the igniter 6 boosts the voltage of the electric current received from the first wires 9 and 9a to excite the second electrode 3 to emit electrons. The electrons emitting from the second electrode 3 further excite the mercury contained in the light tube 2 to discharge electrons, thereby conducting the electric current to the first electrode 3a. In the course of mercury discharge, ultra violet rays are emitted to cause the phosphor coating to generate visible light.

The gas absorbers 4 and 4a are activated gas absorbers formed at the ends of electrodes 3 and 3a connecting to the second wires 5 and 5a to absorb oxygenic gas, such as O<sub>2</sub>, CO, CO<sub>2</sub> and H<sub>2</sub>O.

The gas absorbers made of alloy containing zirconium, vanadium and iron are capable of being activated during the manufacturing process at an activation temperature about 390 degrees Celsius to form the activated gas absorbers 4 and 4a which can absorb oxygenic gas at normal temperature. In other words, no additional expensive heating equipments are required to a ctivate the zirconium-vanadium-iron gas a bsorber to form the activated gas absorbers 4 and 4a at such a higher temperature like 900 degrees Celsius as required for the traditional barium-based gas absorbers. This advantage greatly saves the costs for manufacturing the CCFL and simplifies the manufacturing process, because no such additional components are required. Moreover, the activated gas absorbers 4 and 4a provide stronger oxygenic gas absorption ability and render lower oxidizing rate of second electrode 3 than the traditional ones. The activated gas absorbers 4 and 4a are capable of being made by various processes. For examples, they may be coated with a layer of zirconium-vanadium-iron alloy on the surfaces of electrodes 3 and 3a by means of sputtering and disposition or they may be integrally formed with the electrodes 3 and 3a as an entirety.

The housing 1, enclosing the light tube 2, is attached to the igniter casing 7 for protection of the same. The sealing between the housing 1 and the igniter casing 7 may be air-tight such that the chances air entering from outside of housing 1 into the light tube 2 to fuel the second electrode 3 oxidation is reduce. Accordingly, the housing 1 is able to

keep the light tube 2 warm, as it is functioning. As an alternative, a gas passage may be formed between the interface of the housing 1 and igniter casing 7 in order to balance the pressure within and without the housing 1 and facilitate heat dissipation. The housing 1 may be colored with red, green or blue, the three basic colors, or any other colors, for purposes of colorful illumination. It is noted that the housing 1 can be made of material such as glass or plastic to achieve the above-mentioned effects.

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Referring to FIGs. 2 through 7, variously shaped second electrodes are illustrated. In FIG. 2, the second electrode 3A is shaped as a single layer plate on which the gas absorber 4A is formed. In FIG. 3, as an alternative, the second electrode 3B is shaped as a plate with two layers sandwiching the second wire 5. In FIG. 4, as another alternative, the second electrode 3C is shaped as a rod with the second wire 5 attached to its end. In FIG. 5, as another alternative, the second electrode 3D is made as a tube having a cylindrical side wall defining an inner hollow portion to the end of which the second wire 5 is attached. In FIG. 6, as another alternative, the second electrode 3E is shaped as a spiral that has a constant cross-section along the longitudinal direction to the end of which the second wire 5 is attached. In FIG. 7, as another alternative, the second electrode 3F is shaped as a spiral that has a cross-section varying along the longitudinal direction to the end of which the second wire 5 is attached. The various shapes enlarge the surface area of the second electrode 3 (3A-3F) for better emitting electrons to excite the mercury in the light tube 2. Additionally, the enlarged surface makes the second electrode less susceptible to oxidation than the conventional ones, i.e., wire-shaped second electrodes, due to their rather small surface area. Furthermore, the various shapes enhance the second electrodes' capability to survive an impact force and still keep their structures intact.

Referring to FIGs. 8 through 11, various shapes of light tub are illustrated. In FIG. 8, the light tube 2A is shaped as a spiral with a constant area of cross-section along its longitudinal direction. In FIG. 9, as an alternative, the light tube 2B is shaped as a spiral with a wider bottom tapering vertically to the top. In FIG. 10, as an alternative, the light tube 2C is shaped as a spiral with a wider top tapering vertically to the bottom. In FIG. 11, as another alternative, the light tube 2D is made in a coil-shaped on a plan. The various shapes reduce the space occupied by the light tube 2 (2A-2D), thereby making the CCFL compact. In addition, the above-disclosed housing 1 can be transparent, milky or sand-polished for various effects of illumination.

One skilled in the art will understand that the embodiment of the present invention as shown in the drawings and described above is exemplary only and not intended to be limiting.

It will thus be seen that the objects of the present invention have been fully and effectively accomplished. It embodiments have been shown and described for the purposes of illustrating the functional and structural principles of the present invention and is subject to change without departure from such principles. Therefore, this invention includes all modifications encompassed within the spirit and scope of the following claims.

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